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Avogadro's Number. mole. molar mass. Empirical Formula.
 6.02×10^{23} ; counts extremely small particles. an SI base unit used to measure the quantity of matter. Mass in grams of one mole of any pure substance.

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Formula with smallest whole-number ratio of the elements.

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gas law that states that the rate of effusion is inversely... chapter 10 chemistry test Flashcards and Study Sets | Quizlet the number of representative particles contained in one mole of a substance; equal to 6.02×10^{23} particles Chapter 10 Chemistry

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reaction? (3 marks) 3. Explain four different types of chemical reaction with suitable examples. (8 marks) 4. How do we balance a chemical equation? (4 marks) 5. Why are oil and fat containing food items flushed with nitrogen? (3 marks) 6.

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the correct statement below. a. calorimeter d. temperature b. calorie e. specific heat c. joule f. heat ____ 1. quantity of heat needed to raise the temperature of 1 g of water by 1 C ____ 2. SI unit of energy ____ 3. energy transferred between 2 objects

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because of temperature
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