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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $PE = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r}$ First we consider the interaction between Na^+ ion and

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6 Cl⁻ ions at distance r Applying
Coulomb's law we have $Q_1 Q_2$
 $(-6e)(+e) = -6e$ PE₁ = = = $4\pi \epsilon_0 r$ $4\pi \epsilon_0 r$ $4\pi \epsilon_0 r$
 $4\pi \epsilon_0 r$ Similarly, we now consider 12
Na⁺ ions as second nearest neighbors
at a distance r $Q_1 Q_2 (+12e)(+e) = 12e$
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